Handout 1-01:NOMENCLATURE WORKSHEET

Write a correct name for each of the following compounds.

1. AgI
2. $CuSO_4$
3. NaCN
4. PCl_5
5. $Fe(N0_3)_2$
6. $(NH_4)_2SO_4$
7. RbSCN
8. $AlPO_4$
9. $Hg(NO_3)_2$
10. $BiBr_3$
11. K_2CO_3
12. $CuSO_3$
13. SiF_4
14. $Pb(BrO_2)_2$
15. Na_3PO_4
16. $SnSO_4$
17. SO_3
18. $K_2Cr_2O_7$
19. $Al_2(CO_3)_3$
20. $KHCO_3$
21. $Cu(C_2H_3O_2)_2$
22. HNO_2
23. $Fe(ClO)_2$
24. $HgCr_2O_7$
25. NI_3
26. MnS_2
27. Na_2SO_3
28. H_2SO_3
29. Li_2O
30. CdS
31. $Ni(NO_2)_2$
32. SCl_4
33. $AgBrO_4$
34. Mg_3N_2
35. CaH_2

Write the correct chemical formula for each of the following.

1. Silver chloride
2. Cupric iodide
3. Zinc sulphate
4. Ammonia
5. Strontium fluoride
6. Potassium cyanide
7. Silver sulphide
8. Sulphur trioxide
9. Phosphoric acid
10. Aluminum nitrate
10. Silver hydroxide
12. Ferrous sulphate
13. Cesium perbromate
14. Sodium carbonate
15. Potassium permanganate
16. Strontium bromate
17. Antimony (III) chloride
18. Nitrogen dioxide
19. Zinc acetate
20. Nitrous acid
21. Plumbous iodate
22. Potassium chlorate
22. Chromium(III) oxide
24. Chromium(III) iodide
25. Aluminum oxide
26. Cobaltous chlorate
27. Ammonium bromate
28. Calcium hypobromite
29. Calcium sulphide
30. Calcium phosphite
30. Nitrogen(I) oxide
32. Nitrogen(IV) oxide
33. Strontium hydroxide
34. Calcium nitrite
35. Aluminum hypochlorite
36. Titanium(III) chloride
37. Silver carbonate
38. Ferrous phosphate
39. Barium hypobromite
40. Aluminum acetate
41. Plumbous carbonate
42. Iron(II) bromide
43. Sodium thiocyanate
44. Lithium carbonate
45. Silver nitride
46. Sodium bicarbonate
47. Arsenic(III) chloride
48. Antimonic bromide
49. Cobaltic acetate
50. Hydroiodic acid
51. Aluminum nitride
52. Ammonium carbonate
53. Diphosphorus pentoxide
54. Chromium(III) carbonate
55. Hypobromous acid
56. Nickelous chlorate
57. Silicon dioxide
58. Ammonium iodate
59. Lithium phosphite
60. Dinitrogen pentoxide
61. Cadmium nitrate
62. Lead(IV) iodate
63. Nickel(III) iodide
64. Calcium phosphate
65. Silver chromate
66. Magnesium sulphite
67. Mercurous iodate
68. Stannic oxide
69. Barium carbonate
70. Potassium arsenide