

Strong Acids and Bases, Pop up List

Here is a list of strong acids and bases that completely dissociate in aqueous solutions. Though it is not necessary for this module, it is helpful to memorize the strong acids and bases for chemistry courses at the university level.

Strong Acids

HCl - hydrochloric acid

HNO₃ - nitric acid

H₂SO₄ - sulfuric acid

HBr - hydrobromic acid

HI - hydroiodic acid

HClO₄ - perchloric acid

Strong Bases

LiOH - lithium hydroxide

NaOH - sodium hydroxide

KOH - potassium hydroxide

Ca(OH)₂ - calcium hydroxide

Sr(OH)₂ - strontium hydroxide

Ba(OH)₂ - barium hydroxide