# FROM CRIME SCENE TO

MODULE 1 CHAPTER 7: Over the Counter Pharmaceuticals



### 01. INTRODUCTION

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### 02. TOXICITY

**03. TOX** 

05. CASE REPORTS 06. CANLII CASE STUDY



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### TOXICOKINETICS

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### **PHARMACY DRUG SCHEDULES**



Schedule I drugs: require a prescription for sale and are provided by a pharmacist

Schedule II drugs: less strictly regulated, but still require professional intervention from the pharmacist at the point of sale and possibly referral to a practitioner.

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Schedule III drugs: Although available without a prescription, these drugs are to be sold from the self-selection area of the pharmacy under the **direct supervision** of the pharmacist

**Unscheduled drugs**: can be sold **without professional supervision**. These drugs may be sold from any retail outlet



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### 01.

### • LAW ENFORCEMENT DRUG SCHEDULING

### Drug, substances, and certain chemicals used to make drugs are classified into five distinct categories or schedules depending on:

- the drug's acceptable medical use
- the drug's abuse or dependency potential

**Schedule I:** substances or chemicals with no current accepted medical use and a high potential for abuse E.g., heroin, LSD, ecstasy

**Schedule II:** substances or chemicals with a high potential for abuse, with use potentially leading to severe psychological or physical dependence.

E.g., cocaine, methamphetamine, hydromorphone, oxycodone, fentanyl, Adderall, Ritalin

**Schedule III:** substance or chemicals with a moderate to low potential for physical and psychological dependence. E.g., Tylenol with codeine, ketamine, anabolic steroids, testosterone  $\bigcirc$ 

**Schedule IV:** substances or chemicals with low potential for abuse and low risk of dependence. E.g., Xanax, Valium, Ambien, Tramadol

**Schedule V:** substances or chemicals with lower potential for abuse than Schedule IV and consist of preparations containing limited quantities of certain narcotics. Usually used for antidiarrheal, antitussive, and analgesic purposes E.g., Lomotil, Motofen, Lyrica



### **DRUG SCHED**

	Drug	Туре		
2	Acetylsalicylic acid and its salts	80 mg or less per doses intended for pediatric use		
		Adult use in strengths of 81 mg per dosage unit and 650 mg or g		
		325 mg and 500 mg per dosage unit		
	Acetaminophen	Administration by IV injection		
		Fixed dose combinations containing more than 20,000 mg		
		Sustained release formulations greater than 650 mg per unit or combinations, in package sizes containing 20,000 mg or less		
		Sustained release formulations up to and including 650 mg per units; Immediate release tablets, capsules, suppositories or liquid		
	Ibuprofen or its salts	Except when sold in a dosage form that provides 400 mg or less form that provides 600 mg or less per dosage unit		
		Fixed dose combinations, containing more than 6,000 mg		
		Immediate release containing 400 mg or less per dosage unit, ir that provides 600 mg or less per dosage; Fixed-dose combinatio		
		Immediate release containing 400 mg or less per dosage unit in		

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greater per dosage unit			
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in package sizes of more than 50 units; Fixed-dose	
unit, in package sizes containing no more than 50 d	U
s per dosage unit or in a modified release oral dosage	

n package sizes exceeding 18,000 mg; Modified-release ons, in package sizes containing 6,000 mg or less	
n package sizes of up to 18,000 mg	U

### **ASPIRIN**

### WHAT is it?

### Aspirin, or **acetylsalicylic acid** Is a drug in the family of **salicylates**

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- **General USES**

- used in low-term low-doses to prevent heart
- attacks and cancer



• **analgesic** (against minor pains and aches), • **antipyretic** (against fever)

### • anti-inflammatory

It also has an **anticoagulant** effect and is

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### **HISTORY OF ASPIRIN**



Aspirin was first isolated from **meadowsweet** by German researchers. While the extract was **somewhat effective,** it also caused **digestive problems**, and even **death** when consumed in high doses

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### 1897

German researchers Arthur Eichengrun and Felix Hoffman derivatized one of the hydroxyl functional groups in salicylic acid with an acetyl group, which greatly reduced the negative effects. This was the first synthetic drug, not a copy of something that existed in nature, and the start of the pharmaceutical industry.



### 1853

French chemist Charles Frederic Gerhardt **neutralized** salicylic acid by **buffering** it with **sodium** and **acetyl chloride**, creating **acetylsalicylic anhydride**. His product **worked**, but he had no desire to market it and abandoned his discovery

### ACETAMINOPHEN

### WHAT is it?

### Acetaminophen, or paracetamol, is a **pain reliever** and a **fever reducer**

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### **General USES**

Acetaminophen is used to treat:

• mild to moderate pain • Backache

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- Fever
- Headache
- Muscle aches
- Arthritis

- Toothaches
- Sore throat
- Cold & flu

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### **IBUPROFEN**

### WHAT is it?

Ibuprofen is a **nonsteroidal anti-inflammatory** drug (NSAID) widely marketed under various trademarks including Act-3, Advil, Brufen, Motrin, Nuprin, and Nurofen.



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### **General USES**

- arthritis pain
- Relieves **fever**





### • Treats inflammation such as strain, sprains, and

• Everyday painkiller for a **variety of aches** and

pains including back pain and toothache

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### **COMMONALY MISUSED Over the Counter (OTC) MEDICATIONS**

### **WHAT**

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- Cough suppressant found in many OTC cold medicines
- Most common sources of abused DXM: "extra-strength" cough syrup, tablets and gel capsules
- May be swallowed in its original form, mixed with soda for flavour, called "robo-tripping" or "skittling", or injected
- Often misused in combination with other drugs, such as alcohol and marijuana

### HOW

- An opioid without an effect on pain reduction
- Does not act on opioid receptors Causes a depressant effect in large does, and sometimes a hallucinogenic effect, similar to PCP and ketamine • Repeated use can lead to addiction





### **EFFECTS**

- Short-term effects: range from mild stimulation to alcohol- or marijuana-like intoxication
- At high doses, a person may have hallucinations or feelings of physical distortion, extreme panic, paranoia, anxiety, and aggression
- Misuse of DXM products containing acetaminophen can cause liver damage

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### **ČOMMONLY MISUSED OTC MEDICATIONS**

### **WHAT**

### HOW

- An anti-diarrheal available in tablet, capsule, or liquid form
- Misused by swallowing large quantities



- Opioid designed not to enter the brain
- May act similar to other opioid when taken in large amounts or in combination with other substances

# Loperamide

### **EFFECTS**

- Short term: misused to lessen cravings and withdrawal symptoms
- Can cause euphoria, lead to fainting, stomach pain, constipation, eye changes and loss of consciousness
- May also cause an erratic or rapid heartbeat, and kidney problems
- Effects may increase if combined with other substances

**02. Toxicity** 

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### **DOSING AND EFFECTS**



L'Abbé plots of the proportions of patients improved on high and low doses for the individual direct comparisons with aspirin, ibuprofen and acetaminophen. The size of the circle representing a trial is proportional to the number of patients studied in the trial

Analysis	Aspirin	Ibuprofen	Paracetamol
Number of trials with comparisons	18	20	12
Numerically better with higher dose	12	16	9
Statistical significance with higher dose	2	5	4

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### **ACUTE ACETAMINOPHEN INGESTION**



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- Semilogarithmic plot of **plasma acetaminophen** levels vs time.
- The time coordinates refer to **time after ingestion**.
- represent peak levels.
- The graph should be used only in relation to a **single** acute ingestion.

Rumack-Matthew nomogram for single acute acetaminophen ingestions

• Serum levels drawn **before 4 hours** may not

- The lower solid line 25% below the standard
- nomogram is included to allow for **possible errors** in
- acetaminophen plasma assays and estimated time from ingestion of an overdose.

### **STAGES OF ACUTE ACETAMINOPHEN POISONING**

Stage	Time Post-Ingestion	
I	0-24 hours	Nausea, von
II	24-72 hours	Right upper
III	72-96 hours	Vomiting and Sometimes r
IV	> 5 days	Resolution o organ failure

Mild poisoning **may not** cause symptoms.

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- When present, symptoms of acute acetaminophen poisoning are **usually minor** until **248 hours** after ingestion.
- Symptoms occur in 4 stages, including nausea, vomiting, and right upper quadrant abdominal pain.
- **Renal failure** and **pancreatitis** may occur, occasionally without liver failure.
- After > 5 days, hepatotoxicity resolves or progresses to multiple organ failure, which can be fatal.

Description

niting

quadrant abdominal pain (common)

d symptoms of liver failure renal failure and pancreatitis

of hepatotoxicity or progression to multiple (sometimes fatal)

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### **SALICYLATE POISONING**

### Signs of Aspirin and Other **Salicylate Poisoning**

- Vomiting
- Tinnitus
- Confusion
- Hyperthermia
- Respiratory alkalosis
- Metabolic acidosis
- Multiple organ failure





Significant salicylate toxicity is suggested by serum levels much higher than the **therapeutic range** of 10-20 mg/dL (0.725 to 1.45 mmol/L).

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### **Diagnosis of Aspirin and Other Salicylate Poisoning**

If poisoning is suspected, the salicylate level in **blood** serum is measured, in addition to the urine pH and other medical tests

The serum must be analyzed at least **a few hours** after ingestion; **absorption** is usually almost complete 6 hours following ingestion

### **SALICYLATE POISONING**

### **Adverse Effects of Salicylates**

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- Impair **cellular respiration** by uncoupling **oxidative** phosphorylation
- Stimulate **respiratory centers** in the **medulla**, causing primary respiratory alkalosis, which is often unrecognized in young children
  - Eventually, as salicylates disappear from the **blood**, enter the cells, and poison mitochondria, metabolic acidosis becomes the primary acid-base abnormality
  - Salicylate poisoning also causes **ketosis, fever**, and, even when systemic hypoglycemia is absent, low brain glucose levels



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### Effects of pH

### Low pH

### High pH

Salicylates are **weak** acids that cross cell membranes **relatively** easily; thus, they are more toxic when blood pH is low

**Excretion** of salicylates increases when urine pH increases



### PEDIATRIC FATALITIES ASSOCIATED WITH OTC MEDICATIONS

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## **O3. TOXICOKINETICS**

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### **ABSORPTION OF ASPIRIN**

When ingested orally, acetylsalicylic acid is **rapidly absorbed** in both the **stomach** and **proximal small intestine**.

- The non-ionized acetylsalicylic acid passes through the stomach lining by passive diffusion
- Ideal absorption of salicylate in the stomach occurs in the pH range of 2.15 - 4.10

Peak plasma salicy concentrations oc between 1-2 hou post-administratio

Absorption may vary depending on several factors, including **route, dosage form**, rate of **tablet dissolution,** gastric **contents**, gastric **emptying time,** and gastric **pH**.



- Intestinal absorption of acetylsalicylic acid occurs at a much faster rate
- At least half of the ingested dose is hydrolyzed to salicylic acid in the first-hour post-ingestion by esterases found in the gastrointestinal tract



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Aspirin is distributed to body tissues shortly after administration.



Aspirin is known to cross into the placenta.



The kidney, liver, heart, and lungs are also found to be rich in salicylate concentration after dosing.

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Minimal concentrations are found in feces, bile, and sweat

The plasma contains high levels of salicylate, as well as tissues such as spinal, peritoneal and synovial fluids, and saliva.



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### **METABOLISM OF ASPIRIN**



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**Gentisic acid** 

The major metabolites of acetylsalicylic acid are salicylic acid and salicyluric acid. A small portion is converted to gentisic acid and other hydroxybenzoic acids



Salicyluric acid



### **ELIMINATION OF ASPIRIN**



• Excretion of aspirin occurs mainly through the kidney in the form of free salicylic acid, salicyluric acid, and phenolic and acyl glucuronides

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### Half-life

- The half-life of ASA in the circulation ranges from **13 19 minutes**.
- Blood concentrations drop rapidly after complete absorption.
- The half-life of the ASA ranges between **3.5 and 4.5 hours**



The rate of unmetabolized acetylsalicylic acid (ASA) is often **variable**, ranging from **10% to 85%** in the urine, and heavily depends on **urinary pH.** Acidic urine generally aids in reabsorption of ASA by the renal tubules, while alkaline urine increases excretion

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### **ABSORPTION OF IBUPROFEN**



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Absorbed well orally. The peak serum concentration can be attained in 1 to 2 hours



When ibuprofen is administered immediately after a meal there is a slight reduction in the absorption rate, but there is no change in the extent of the absorption



When orally administered, the absorption of ibuprofen in adults is very rapidly done in the upper GI tract



Average AUC: 70 mcg\*h/ml

### **DISTRIBUTION OF IBUPROFEN**

### **VOLUME OF DISTRIBUTION**

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The volume of distribution of ibuprofen is of O.1L/kg

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### **PROTEIN BINDING**

Ibuprofen dosage is more than 99% bound to plasma proteins and site II of purified albumin. Binding appears to be saturable and becomes non-linear at concentrations exceeding 20 mcg/ml.

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### carboxy-ibuprofen

### **ELIMINATION OF IBUPROFEN**

90% of the administered dose is eliminated in the urine

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**99%** of the administered dose is excreted as metabolites. The other 1% is unchanged drug.



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The clearance rate ranges between 3-13 L/h depending on the route of administration and dosage Ibuprofen is completely eliminated 24 hours after the last dose



The **serum** half-life of ibuprofen is **1.2-2 hours** In patients with compromised liver function, the half-life can be **prolonged to 3.1-3.4 hours** 

# **04. BIOLOGICAL FLUID TESTING**

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### **SALICYLATES IN BIOLOGICAL FLUIDS**

Therapeutic range of salicylates: 150-300 mg/L

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Patients are often symptomatic at salicylate concentrations higher than 400-500 mg/L Patients with salicylate concentrations approaching or exceeding 1000 mg/L usually have serious or life threatening toxicity

Peak serum concentration may not occur for 4-6 hours, so concentrations obtained before that time may not reflect peak levels

Salicylate levels are typically monitored in the **blood** and the urine

The Trinder spot test is a presumptive test to detect salicylates in the urine The reagent is mixed with urine, and detects **salicylic acid** - a metabolite of salicylates Solution is a composed of mercuric chloride, water, hydrochloric acid, and ferric nitrate

- Blue or purple = positive
- No change = **negative**
- **Brown** = false positive caused by the presence of phenothiazines

94% sensitivity and 74% specificity for identifying patients whose salicylate concentrations are greater than 300 mg/L





**Positive Result** 

**False Positive Result** 

Confirmatory testing is done by HPLC

### **ACETAMINOPHEN IN BIOLOGICAL FLUIDS**

Therapeutic concentration: 10-15 mg/L Toxic concentration: 100-150 mg/L Comatose-fatal concentration: 200-300 mg/L

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Presumptive Colorimetric Test for Acetaminophen

- Performed on urine or protein-free filtrate of **blood**
- Hydrochloric acid is added and the solution is heated to 100°C
- A **blue colour** after the addition of 1% o-cresol in water and ammonium hydroxide constitutes a positive test for acetaminophen. **Therapeutic** or **toxic** use of acetaminophen can be identified using this colour reaction





### **Confirmatory testing is** done using HPLC

# **05. CASE REPORTS**

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### **CASE REPORT - IBUPROFEN OVERDOSE**

### **OVERVIEW**

A 26-year-old female deliberately ingested up to **132 tablets** of **800 mg** sustained-release **ibuprofen**, equivalent to approximately **105 g** 

Despite gut decontamination with multidose activated charcoal and correction of the metabolic acidosis with sodium bicarbonate and haemofiltration, the patient did not survive.

### **TOXICOLOGY SCREENING**

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Post mortem samples of whole **blood**, **urine**, **gastric contents** and **liver** extract were analysed at the local toxicology laboratory for ibuprofen and other drugs.

Ibuprofen concentrations were measured by **high-pressure liquid chromatography** with **ultraviolet detection**.

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### RESULTS

Antemortem serum ibuprofen concentrations were 760 mg/L on arrival, rising to a peak concentration of 1,050 mg/L 90 minutes later.

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### **CASE REPORT - IBUPROFEN OVERDOSE**

In Blood...

Therapeutic dose: 15-30 mg/L

Toxic dose: >200 mg/L

Comatose-Fatal: >350 mg/L

No other drugs were detected in a broad toxicology screen; analysis of the antemortem and postmortem serum samples only detected atracurium and lignocaine given following admission to the hospital.



### Postmortem ibuprofen concentrations:

- 518 mg/l whole blood
- 264 mg/l urine
- 116 mg/l gastric contents
- 74 mg/kg liver extract

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### **CHILD ACETAMINOPHEN OVERDOSE**

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22-day-old male admitted to the emergency room after realising that an acuteacetaminophen overdose had occurred,following a routine procedure

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The parents had been instructed to give him **40 mg** of acetaminophen before the procedure. The infant's weight was **4.1 kg**, so this was a recommended dose of **10 mg/kg**.



However, the infant had mistakenly been given about **800 mg**, or **200 mg/kg**, of acetaminophen by his parents before the procedure. The bottle of acetaminophen showed a concentration of **80 mg/mL**, which was **misinterpreted by the parents**, in that they believed that the bottle contained **80 mg of acetaminophen in total** 





The child was given **10 mL**, or about **half the bottle**, with the intent of giving him **40 mg** 



After the procedure, the physician instructed the parents to give him another dose of acetaminophen if he seemed uncomfortable. At that point, the mother commented that "it **seemed like a lot of medicine**" and the error was discovered O

### CHILD ACETAMINOPHEN OVERDOSE



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The acetaminophen blood concentration drawn four hours after the overdose was substantially elevated at **1243 umol/L.** (upper end of **therapeutic range 66-199 umol/L**)



Given that the patient had received more than the **toxic dose of 150 mg/kg** and because the four-hour blood concentration level of acetaminophen was in the probable toxicity tange on the Rumack-Matthev nomogram, treatment with **N-acetylcysteine** was recommended



N-acetylcysteine **reduces the hepatotoxic effects** of acetaminophen overdose by **replenishing glutathione stores**, thereby enhancing production of the **nontoxic metabolites** 



The infant was sent home after 48 hours, remained clinically well, and did not show evidence of long-term consequences of the accidental overdose



# **06. CANLII CASE STUDY**

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### E.D. v S.K., 2017

### BACKGROUND

On February 2, 2014, a 20 year old woman was brought to the hospital after she was found at her apartment struggling to breathe and with decreased consciousness

The patient had a medical history that included **idiopathic intracranial hypertension**, was on medications including acetazolamide, and had been prescribed **Tylenol 3** and **222s** (contain **ASA**, caffeine, & codeine)



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The paramedics noted an almost **empty container of 222s** and thought the patient had overdosed. However the patient's mother advised health professionals that it was not an overdose and informed them of the patient's medical history



From the morning of February 3, 2014 and over the next few days, the diagnosis of cerebral death was confirmed. Life support was removed, and the patient passed away on February 7,2014

**Issue:** The applicant, the patient's mother, expresses concern that the respondent, a physician, failed to provide appropriate medical care to her daughter



### **TOXICOLOGICAL FINDINGS**

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### **Initial Results:**

Arterial **blood** gas showed severe metabolic acidosis, salicylate at a therapeutic level, and opiates were found on urine toxicology, consistent with the patient's medication

### **Opinion accidental poisoning by of Dr. M:**

- Low levels of ASA/acetazolamide can be toxic if the use is chronic and regular
- ASA levels could be low in the plasma but high in the tissues and able to cross the blood-brain barrier when a patient is acidotic
- This effect is **amplified by acetazolamide** and explains why the ASA would be **undetectable** on the toxicology screen
- The patient's **brain was vulnerable** as she had idiopathic intracranial hypertension, causing high pressure in the brain and was on medications which had unforeseen dangerous side effects, even at therapeutic doses
- Concluded the patient suffered from accidental poisoning from chronic use of medication at normal doses.

### E.D. v S.K., 2017

### DECISION



The medical care provided was deemed acceptable, and the Board confirmed the committee's decision to take no action against the respondent

# **07. LIST OF REFERENCES**

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